

## Physics 10154 - Exam #5A

Points will be deducted if you don't show your work (or if some parts are incorrect) even if you get the right answer. Clearly indicate your answer with a circle or box and remember to include correct units and significant figures.

1. (30 pts) A 12.0-kg rock specimen is estimated to have a density of  $2240 \text{ kg/m}^3$ .

a) If this rock is immersed in water, how long will it take (in seconds) to fall 3.00 meters to the bottom of a container?

b) While immersed in water, the rock is placed on a scale to measure its apparent weight. What weight shows on the scale?

a)

$$V_0 = \frac{m_0}{\rho_0} = .00536 \text{ m}^3$$

$$\Sigma F_y = F_g - F_B = ma$$

$$mg - \rho_f V_0 g = ma$$

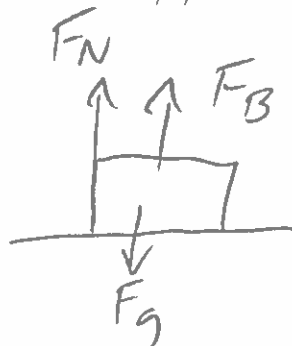
$$3.0 = 0 + \frac{1}{2}(5.425)t^2$$

$$117.6 - 52.5 = 12a$$

$$\boxed{t = 1.05 \text{ s}}$$

$$a = 5.425 \text{ m/s}^2$$

b) Apparent weight =  $F_g - F_B = \boxed{65.1 \text{ N}}$



$$\text{or } \Sigma F_y: F_N + F_B - F_g = 0$$

scale reading  $\rightarrow F_N = F_g - F_B$

or 6.64 kg

2. (30 pts) 56 grams of ice at a temperature of  $-25^{\circ}\text{C}$  is dropped into a 230-gram aluminum container which has an initial temperature of  $65^{\circ}\text{C}$ .

What is the final temperature of the system? If the final temperature of the system is 0, how much ice melts?

The specific heat of aluminum is  $900 \text{ J/kg } ^{\circ}\text{C}$ . Other constants needed are on your formula sheet.

To melt ice:

$$\Delta Q = (.056)(2090)(25) + (.056)(333000) \\ = 21574 \text{ J}$$

To cool Al:

$$\Delta Q = (.230)(900)(-65) = -13455$$

not all ice melts,  $T_F = 0^{\circ}\text{C}$

$$\Delta Q_{\text{ice}} + \Delta Q_{\text{Al}} = 0$$

$$2926 + m(333000) - 13455 = 0$$

$$m = \frac{10529}{333000} = \boxed{.032 \text{ kg}}$$

3. (20 pts) A double-paned square window measures 34 cm on a side. The window has two panes of glass (0.75 cm thick for each pane) that sandwich a 0.50 cm layer of air between them. If the temperature outside is 42°F and the temperature inside is 68°F, determine how much money is spent to replace the heat energy lost by the window in one day.

Assume the thermal conductivity of glass is 0.85 W/m K and the thermal conductivity of air is 0.0350. Assume the cost of energy is 12 cents per kilowatt-hour. Answer to the nearest cent.

$$R\text{-value} = \frac{l_1}{k_1} + \frac{l_2}{k_2} + \frac{l_3}{k_3}$$

$$\frac{.0075}{.85} + \frac{.0050}{.0350} + \frac{.0075}{.85} = 0.1605$$

$$P = \frac{A \Delta T}{R\text{-value}} \quad \Delta T = 26^\circ\text{F} = 14.4^\circ\text{C}$$

$$P = \frac{(.34)^2 (14.4)}{0.160} = 10.4 \text{ W}$$

$$E = P \cdot t = (10.4361)(86400 \text{ s})$$

$$= 9.0168 \times 10^5 \text{ J}$$

$$= 0.250 \text{ kW}\cdot\text{hr}, \quad \frac{12 \text{¢}}{\text{kW}\cdot\text{hr}}$$

$$= \boxed{3 \text{¢}}$$

4. (20 pts) If you move 3.0 times further away from a source of sound (assume it radiates uniformly in all direction), by how many dB does the loudness change?

$$\frac{I_1}{I_2} = \frac{1}{9} \quad \text{since } I \propto \frac{1}{r^2}$$

$$\Delta B = 10 \log \left( \frac{1}{9} \right) = \boxed{-9.5 \text{ dB}}$$