## Physics 10164 - Spring 2019 Exam 4B

Partial credit will be given provided you show all work and are solving parts of the problem correctly. Points will be deducted
if you don't show your work even if you get the right answer. Clearly indicate your answer with a circle or a box and remember to include correct units and significant figures.

1. (25 pts) A light source illuminates two parallel slits separated by a distance d. One point on the screen where two rays meet is the 3 rd minimum, counted relative to the central bright fringe. The difference in distances travelled by the two rays is 855 nm (ray 2 travels a little further than ray 1).
a) What is the wavelength (in nm ) of the incident light?
b) If the slit separation $d=5.50 \times 10^{-6} \mathrm{~m}$, what angle $\theta$ do the rays make with the horizontal?

2. (25 pts) A diffraction grating contains 2200 lines/cm. When used with light of a certain wavelength, a third-order maximum is formed at an angle of $25^{\circ}$.
a) What is the wavelength of this light?
b) How many complete orders of visible light (400-700 nm) are visible upon reflecting from this diffraction grating?
3. (25 pts) A hydrogen atom's electron is in energy level $\mathrm{n}=$ 5 , and a photon is emitted by the electron. Determine (a) the energy and wavelength of the photon that has the shortest possible wavelength. Also, determine (b) the energy and wavelength of the photon that has the longest possible wavelength. Answer this question with 3 SF .
4. (25 pts) One possible Uranium fission reaction is
${ }^{1} \mathrm{n}+{ }^{235} \mathrm{U} \rightarrow{ }^{94} \mathrm{Zr}+{ }^{139} \mathrm{Te}+3{ }^{1} \mathrm{n}+$ energy
We will assume the masses of these nuclei are:
neutron ( $\left.{ }^{1} \mathrm{n}\right)=1.008665 \mathrm{u}$
Uranium $\left({ }^{235} \mathrm{U}\right)=235.043924 \mathrm{u}$
Zirconium $\left({ }^{94} \mathrm{Zr}\right)=93.906315 \mathrm{u}$
Tellurium $\left({ }^{139} \mathrm{Te}\right)=138.934730 \mathrm{u}$
a) How much energy is produced by this reaction (in MeV)?
b) How much mass (in kg ) of Uranium would be necessary to power a house that uses $2200 \mathrm{kw}-\mathrm{hr}$ of energy per month, assuming this is the reaction that provides the power?
